

Vitamin B₁₂, a Catalyst in the Synthesis of Prostaglandins

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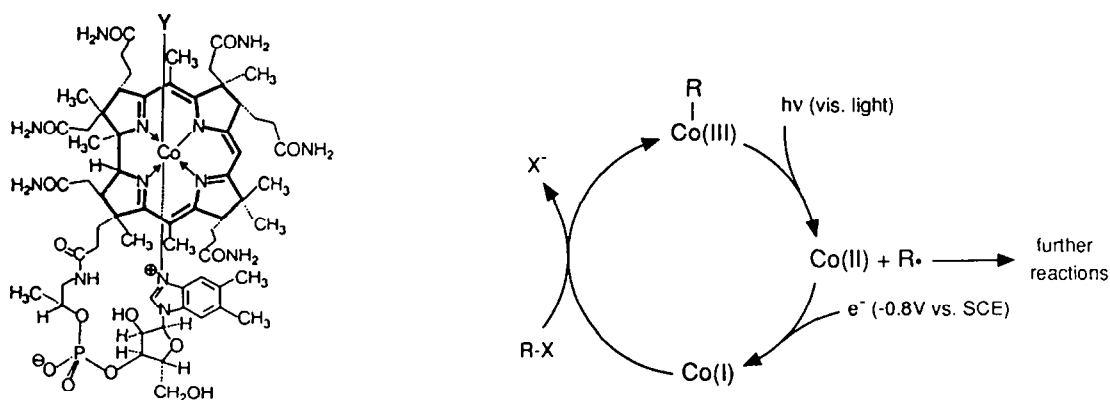
Dedicated to Professor Wang Yu on the occasion of his 80th birthday

(Received in Japan 2 November 1989)

Abstract. - A prostaglandin F_{2α} precursor containing all structural features from C₆ to C₂₀ with 8*R*, 9*S*, 11*R* and 12*R* chirality is obtained by the one step formation of two C-C bonds in the B₁₂-catalyzed radical cyclization-addition sequence starting from a chiral cyclopentene bromoacetal and 1-octyne-3-one. The B₁₂-catalyzed radical cyclization-elimination sequence of a chiral cyclopentene precursor leads to (-)-(3*aR*,6*aS*)-3,3*a*,6,6*a*-tetrahydro-2*H*-cyclopenta[*b*]furan-2-one. Its (+)-(3*aS*,6*aR*)-enantiomer is obtained *via* B₁₂-catalyzed, enantioselective isomerization of cyclopentene oxide to (*R*)-2-cyclopentene-1-ol followed by the B₁₂-catalyzed radical cyclization-elimination sequence of its bromoacetal.

Introduction. - Recent years have seen an impressive development of radical chemistry directed towards organic synthesis¹). The commonly used reagents are organotin compounds, in particular tri-*n*-butyltin hydride²). Mainly for ecological, but also for economic reasons these compounds should be replaced by less toxic and more specifically acting reagents or catalysts. A salient candidate is vitamin B₁₂³), a coenzyme known to promote a series of biological transformations *via* radical intermediates⁴). Since B₁₂ is produced industrially in large-scale fermentation processes⁵), it is commercially available at a reasonable price. B₁₂ and related Co-complexes have been applied as catalysts for various reaction types such as: oxidations, hydrogenations, reductions of functional groups, reductive eliminations, rearrangements and reductive C-C bond formations⁶). Of special interest in organic synthesis is the B₁₂-catalyzed, photochemically induced radical C-C bond formation⁷). The mode of action of these catalysts is known to some extent⁸) (Scheme 1).

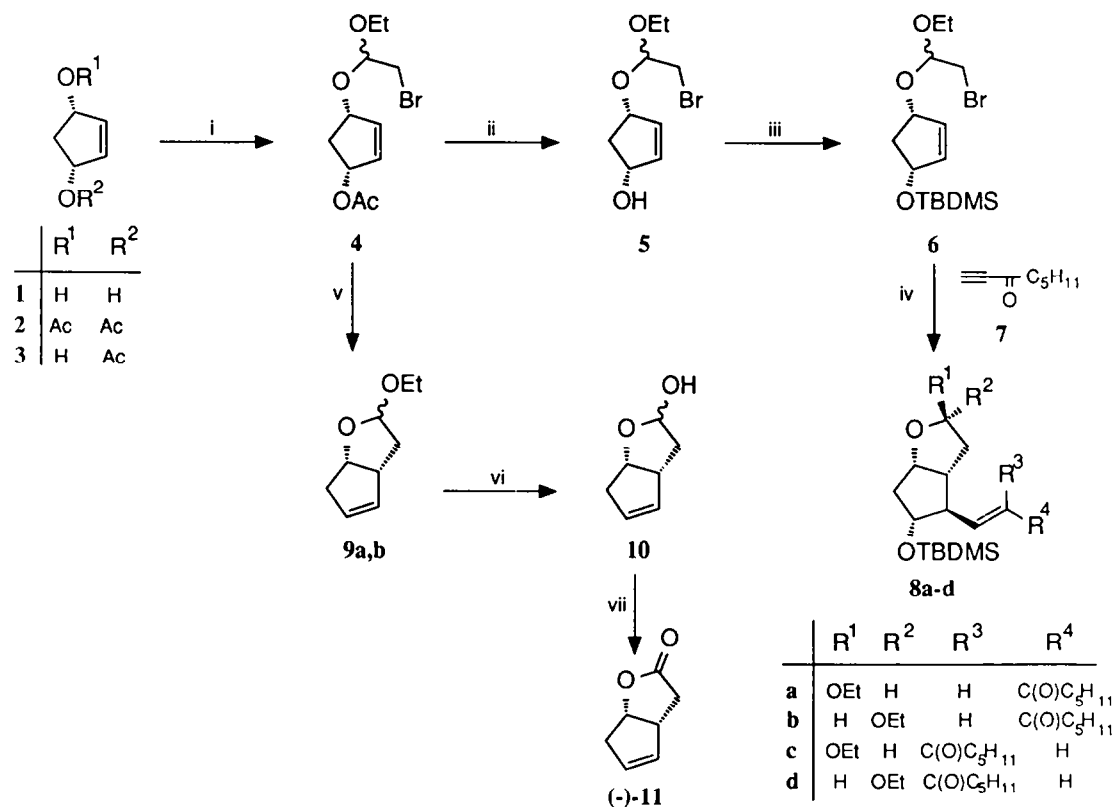
Scheme 1. Generation of Carbon Centered Free Radicals by B₁₂-Photo-Electro-Catalysis



The catalytically active species is cob(I)alamin (B_{12s}), which is obtained by reduction of hydroxocob(III)alamin (B_{12a}) in two consecutive one-electron reductions. The first reduction to cob(II)alamin (B_{12r}) occurs at *ca.* 0 V (*vs.* SCE), the second to B_{12s} at *ca.* -0.8 V⁹. B_{12s} is (different from B_{12r}) a highly reactive species and undergoes rapid bimolecular nucleophilic substitution with primary alkyl bromides RX to form organocob(III)alamins¹⁰. Photochemically induced Co-C bond homolysis with visible light ($\lambda = 400\text{-}600\text{ nm}$)¹¹ affords the persistent radical B_{12r}^{12} and the transient carbon radical $R\cdot$ ¹². B_{12r} is recycled to B_{12s} by a one-electron reduction, whilst the carbon radical undergoes typical "free radical" reactions, *e.g.* addition to olefins¹.

Synthesis of PGF_{2 α} -precursors¹³. - Numerous approaches in PG-syntheses start from chiral cyclopentane precursors bearing functionalities ready to be elaborated into side chains¹⁴. The key feature of two recent reports on PGF_{2 α} -syntheses from *Stork et al.*¹⁵ and *Keck and Burnett*¹⁶ is the cyclization of free radicals generated from a cyclopentene iodoacetal with trapping of the intermediate secondary carbon radical by a precursor of the C₁₃-C₂₀ side chain in a single step. Both cyclization-trapping sequences^{1a,17} are based on radical generation using organotin compounds.

Scheme 2. B₁₂-Catalyzed Synthesis of PG-Precursors from Optically Active Cyclopentene Diol Derivatives



i) 3 + ethylvinylether, NBS, CH₂Cl₂, -40° to r.t., 16h (88%); ii) KOH, 90% EtOH, r.t. (83%); iii) TBDMSCl, imidazole, DMF, r.t. (98%); or i to iii without isolation of intermediates in 96%; iv) 6 equiv. 7, 0.018 equiv. B_{12a} , electrolysis in 0.3M LiClO₄/DMF at -0.9V (*vs.* SCE), vis. light (>55% **8** as a mixture of four isomers (GC)); cis-trans-equilibration: l₂, hexane, hv -> trans isomers **8a,b** (47%); v) 0.01 equiv. B_{12a} , Zn, NH₄Cl/EtOH, r.t., 6h (73%); vi) 1.5% HCl/THF, r.t., 48h (37%, not optimized); vii) CrO₃, acetone/H₂SO₄/H₂O (63%, *e.e.*>98%, not optimized).

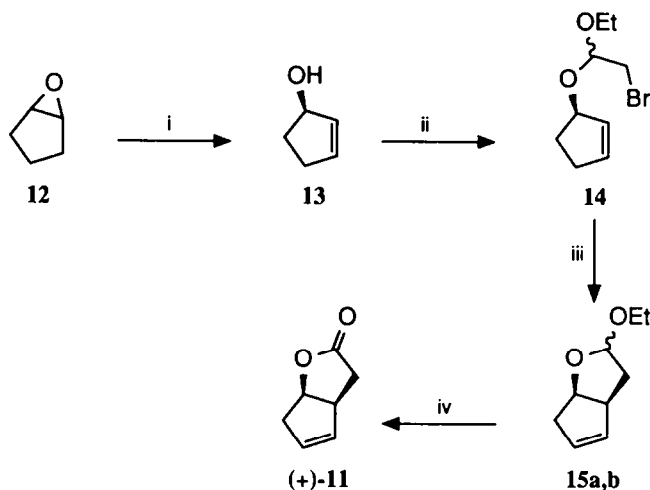
Our synthesis of PGF_{2α}-precursors relies on the original work on B₁₂-catalyzed, reductive radical cyclization¹⁸, *Tada's* report concerning the cobaloxim-mediated synthesis of γ-lactones¹⁹ and the knowledge of the B₁₂-catalyzed radical addition to activated alkynes⁷ (Scheme 2).

The *cis*-diol **1** was obtained by a slight modification of the procedure of *Kaneko et al.*²⁰ starting from cyclopentadiene. Acetylation²¹ gave the diacetate **2**, which was hydrolyzed enantioselectively with porcine pancreas lipase according to *Laumen and Schneider*²² to afford (+)-**3** (e.e. > 98%) in 37% yield from cyclopentadiene. Reaction with ethylvinyl ether and N-bromosuccinimide (NBS)^{15,19} gave the bromoacetal **4**. Since the acetoxy function in **4** is a potential leaving group, it was replaced by a silyloxy function *via* basic hydrolysis to **5** and silylation with *tert.*-butyldimethylchlorosilane (TBDMS-Cl)²³ affording **6** in 72% yield with respect to **3**. The key-step of the synthesis is the reductive B₁₂-catalyzed radical cyclization of **6** with concomitant trapping of the intermediate secondary radical by 1-octyne-3-one **7**²⁴, affording **8** in one operation. The electrolysis of **6** in the presence of a fivefold excess of **7** was performed in a specially designed H-type electrochemical cell²⁵ under irradiation with visible light at a constant cathode potential of -0.9V (vs. SCE) and in the presence of 0.018 equiv. of B_{12a} (with respect to **6**) in 0.3 M LiClO₄/DMF. Under these conditions a mixture of the four diastereomers of **8** (differing in the configuration at the double bond and of the acetal function) was obtained in *ca.* 55% yield. All four isomers were separated by chromatography and characterized individually. Quantitative isomerization to the *trans*-olefins **8a,b** was achieved by exposing a solution of the mixture of diastereomers in hexane, containing some iodine, to visible light. **8a** and **8b** were separated by chromatography and isolated as pure materials in 26% and 21% yield, respectively, with respect to **6**. The two diastereomeric acetals easily underwent acid-catalyzed interconversion. The mixture **8a,b** is an advanced intermediate, from which PGF_{2α} had been obtained in three further steps¹⁵.

The B₁₂-catalyzed cyclization-addition step **6** + **7** → **8** needs some comments: in the absence of either B₁₂ or light, no product **8** is formed at -0.9V; at a more negative potential or with Zn/NH₄Cl as a chemical reducing agent, **7** is reduced to 1-octene-3-one, which traps the intermediate radical affording products lacking the Δ^{13,14} double bond²⁶; as a by-product of the described synthesis 1,3,5-trihexanoylbenzene (the product of B₁₂-catalyzed radical trimerization of **7**) was formed in *ca.* 12% with respect to the total amount of **7**.

B₁₂-catalyzed electrolysis of the acetoxy bromoacetal **4** in DMF at -1.0V afforded the diastereomeric acetals **9a,b** as products of a cyclization-elimination sequence. The same compounds were obtained in 73% yield on treatment of **4** in ethanol with Zn/NH₄Cl in the presence of 0.01 equiv. B_{12a}. Acid hydrolysis gave the crystalline hemiacetal **10**, which on *Jones*-oxidation afforded the known²⁷ lactone (-)-**11** (m.p. 44-45°, e.e. > 98%), an important intermediate in PG-syntheses²⁸.

The enantiomer (+)-**11** was prepared in five steps from cyclopentene oxide **12** (Scheme 3). The achiral epoxide **12** was first transformed to (*R*)-2-cyclopentene-1-ol **13** (67% yield, 62% e.e.) by the recently discovered enantioselective isomerization with catalytic amounts of B_{12a} (0.01 equiv.) and Zn in methanol²⁹. **14** was then obtained quantitatively in the usual way. The B₁₂-catalyzed cyclization of **14** with NaBH₄ in ethanol at 60° afforded the diastereomeric acetals **15a,b** in 71% yield. *Jones*-oxidation followed by crystallization finally gave (+)-**11**³⁰ in 45% yield and an e.e. > 99%. The straightforward transformation of the achiral epoxide **12** to essentially pure (+)-**11** illustrates the power and versatility of vitamin B₁₂ as catalyst in organic synthesis.

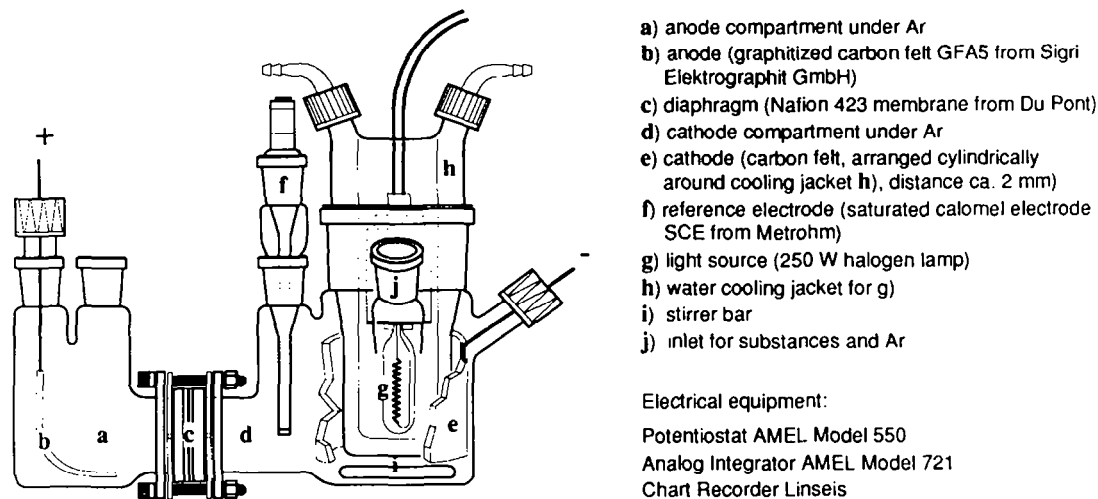
Scheme 3. Synthesis of Optically Active Cyclopenta[b]furans by Enantioselective B₁₂-Catalysis

i) 0.01 equiv. B_{12a}, Zn/NH₄Cl, MeOH, r.t., 8d (67%, e.e.=62%); ii) ethylvinylether, NBS, CH₂Cl₂, -40° to r.t., 16h (100%); iii) 0.035 equiv. B_{12a}, NaBH₄, EtOH, 60°, 70min. (71%); iv) CrO₃, acetone/H₂SO₄/H₂O, 0°, 1h, then crystallization and 2 x recrystallization (45%, e.e.>99%).

Experimental Section

General: Vitamin B_{12a} (hydroxocobalamine hydrochloride, pyrogen-free *Fr. Ph. BP*, 10.7% loss on drying, < 2% cyanocobalamine) from *Roussel Uclaf*, rose bengal (*standard*), thiourea (*purum*), pyridine (*puriss.*), acetic anhydride (*puriss.*), porcine pancreas lipase PPL (cat. no. 62300), N-bromosuccinimide (*purum*), *tert.*-butyldimethylchlorosilane (*purum*), imidazole (*puriss.*), 1-octyne-3-ol (*purum*), LiClO₄ (*purum*) and *p*-toluenesulfonic acid monohydrate (*pract.*) from *Fluka*, CrO₃ (*p.a.*) from *Merck*, NH₄Cl (*purum*), HOAc (98%), H₂SO₄ (*puriss.* 95%) and THF (*purum*) from *Siegfried* and O₂ from *Carba* were used as purchased. CH₂Cl₂ from *Fluka* was distilled over P₂O₅. Cyclopentadiene was freshly distilled from its dimer (*pract.*) from *Fluka* over a 30 cm vigreux column. 4-Dimethylaminopyridine (*purum*) from *Fluka* was recrystallized from ethylacetate. Ethylvinylether (*purum*) from *Fluka* was freshly distilled over sodium. Cyclopentene oxide from *Merck* was distilled, b.p. 98-99°/715 mbar (important: the oxide should be free of chlorinated solvents). 0.1M P-buffers were prepared from Na₂HPO₄•H₂O (*puriss.*) from *Fluka* and NaH₂PO₄•12H₂O (*p.a.*) from *Merck*. Zn-wool (*puriss.*) from *Siegfried* was activated before use: Zn (*ca.* 1 g) in 2N HCl (*ca.* 10 ml) was stirred for 15 min at r.t., followed by washing with H₂O, MeOH and Et₂O (3 x 10 ml each). DMF (*puriss.*) from *Fluka* was filtered through basic Alox from *Merck* before use, except for the electrochemical experiment yielding PG-precursor **8**, where best results were obtained with dry DMF (either *puriss. absolute*, < 0.01% H₂O from *Fluka*, filtered through basic Alox, or distillation of *puriss.* from *Fluka* over P₂O₅, filtered through basic Alox). The electrochemical experiment was carried out in an H-type cell under potentiostatic conditions (see *Fig.*). GC: *Hewlett-Packard* 5890 gas chromatograph, 20 m *Duran* glass cap. column coated with *SE-54*, temp. program from 40° to 200° at a rate of 3°/min. ¹H-NMR: *Varian EM 360L* (60 MHz) and *Bruker AM-400WB* (400 MHz), TMS (= 0 ppm) as internal standard. IR: *Perkin-Elmer* 782 infrared spectrometer. MS: *Varian MAT CH-7A* mass spectrometer, ionization energy 70 eV. UV: *Hewlett-Packard* 8451A diode array spectrophotometer. [α]_D: *Perkin-Elmer* 241 polarimeter. n_D: *Carl Zeiss* refractometer. Melting points: *Büchi* 510. Elemental analysis: performed by

Org.-chem. Mikrolabor ETH, Zürich. TLC: precoated plates, silica gel 60 F₂₅₄ from Merck developed with H₂SO₄/vanilline. Flash chromatography: silica gel for flash chromatography from Baker. HPLC: silica gel 7 μm, column diameter 23 mm, column length 25 cm.



- a) anode compartment under Ar
- b) anode (graphitized carbon felt GFA5 from Sigr Elektrographit GmbH)
- c) diaphragm (Nafion 423 membrane from Du Pont)
- d) cathode compartment under Ar
- e) cathode (carbon felt, arranged cylindrically around cooling jacket h, distance ca. 2 mm)
- f) reference electrode (saturated calomel electrode SCE from Metrohm)
- g) light source (250 W halogen lamp)
- h) water cooling jacket for g)
- i) stirrer bar
- j) inlet for substances and Ar

Electrical equipment:

- Potentiostat AMEL Model 550
- Analog Integrator AMEL Model 721
- Chart Recorder Linseis

Fig. Equipment for Photo Electro Catalysis
(conceived by Ch. Weymuth²⁵⁾)

cis-4-Cyclopentene-1,3-diol (1). A solution of cyclopentadiene (16.0 ml, 193.6 mmol), thiourea (10.0 g, 131.4 mmol) and rose bengal (300 mg, 0.3 mmol) in distilled methanol (1600 ml) was purged with oxygen and irradiated with a 100 W halogen lamp at r.t. for 8 h. After stirring for another 12 h without light, the obtained pink suspension was concentrated *in vacuo* to ca. 100 ml and filtered through *Celite*. Methanol (50 ml) was added to the filtrate and the newly formed suspension filtered again through *Celite* after standing at r.t. for 12 h. The solvent was removed and distillation (100° - 130° (bath temp.)/2•10⁻² mbar) afforded **1** (11.20 g, 58%) as a pale yellow liquid, which solidified upon standing. It was used without further purification. ¹H-NMR (60 MHz, CDCl₃): 1.56 (*dt*, *J* = 15 and 3, HC(2)); 2.73 (*m*, HC(2)); 3.93 (*s*, OH); 4.68 (*m*, HC(1), HC(3)); 6.05 (*s*, HC(4), HC(5)). IR (KBr): 3280_s, 3060_s, 2975_s, 2935_m, 2875_s, 2250_m, 2230_m, 1445_m, 1350_s, 1315_s, 1290_m, 1270_m, 1165_m, 1120_s, 1090_s, 1065_s, 1015_s, 995_s, 950_m, 880_m, 845_m, 780_s, 760_s.

cis-4-Cyclopentene-1,3-diol diacetate (2). To an ice cooled solution of diol **1** (10.01 g, 0.1 mol) and 4-dimethylaminopyridine (1.25 g, 10.0 mmol) in pyridine (100 ml) under N₂ was added acetic anhydride (37.74 ml, 0.4 mol) over 45 min. After stirring for 30 min at r.t. the solution was poured into ice water (200 ml) and extracted with Et₂O (3 x 100 ml). The org. soln. was washed with 10% HCl (2 x 50 ml), 5% NaHCO₃ (2 x 50 ml) and brine (3 x 50 ml), dried (Na₂SO₄) and the solvent removed. Bulb-to-bulb distillation (105° (oven temp.)/2•10⁻² mbar) of the residue afforded **2** (14.57 g, 79%) as a pale yellow liquid. ¹H-NMR (60 MHz, CDCl₃): 1.73 (*dt*, *J* = 15 and 4, HC(2)); 2.05 (*s*, CH₃); 2.90 (*m*, HC(2)); 5.59 (*dd*, *J* = 4 and 7.5, HC(1), HC(3)); 6.13 (*s*, HC(4), HC(5)). IR (neat): 2950_w, 1740_s, 1435_w, 1365_m, 1260_s, 1180_w, 1075_m, 1020_m, 990_w, 960_w, 910_w, 770_w.

(1R,4S)-4-Cyclopentene-1,3-diol monoacetate (3). To an emulsion of diacetate **2** (3.26 g, 17.7 mmol) in 0.1M P-buffer of pH = 7.00 (70 ml) was added a suspension of porcine pancreas lipase PPL (4.00 g) in the same buffer (30 ml). Whilst stirring at r.t., the pH was kept at 7.0 by continuous addition of 1M NaOH. After 48 h 19.12 ml

base had been consumed. The suspension was continuously extracted in a *Kutscher-Stuedel* extractor with Et₂O (150 ml) for 24 h. The org. soln. was dried (MgSO₄) and the solvent removed. The remaining white solid was recrystallized from Et₂O/low boiling petroleum ether to yield **3** (2.02 g, 80%) as colorless needles. M.p. 49° - 51°. ¹H-NMR (60 MHz, CDCl₃): 1.65 (*dt*, *J* = 15 and 4, HC(2)); 2.07 (*s*, CH₃); 2.84 (*m*, HC(2)); 2.63 (*s*, OH); 4.77 (*m*, HC(3)); 5.53 (*m*, HC(1)); 6.07 (*m*, HC(4), HC(5)). IR (KBr): 3390*m*, 2920*w*, 2900*w*, 1825*w*, 1730*s*, 1445*m*, 1415*m*, 1380*m*, 1360*s*, 1325*m*, 1255*s*, 1180*m*, 1085*s*, 1060*s*, 1020*s*, 980*m*, 970*m*, 910*w*, 880*w*, 840*w*, 790*m*. [α]_D²⁰: + 68°·0 (c = 1.64, CHCl₃; e.e.>98% by comparison with an authentic sample (e.e.>99%) kindly submitted by *M. Schneider* and *K. Laumen.*). Higher concentrations of substrate and enzyme gave considerably shorter reaction times, but also lower yields (60% - 70%).

(1*R*,4*S*)-4-(1-Ethoxy-2-bromoethoxy)-2-cyclopentene-1-ol acetate (**4**). To a suspension of alcohol **3** (1.30 g, 9.1 mmol) and N-bromosuccinimide (1.87 g, 10.5 mmol) in dry CH₂Cl₂ (20 ml) under N₂ at -40° was added ethylvinylether (1.24 ml, 13 mmol) dissolved in CH₂Cl₂ (3 ml) over 30 min. After 2 h the solution was allowed to warm to r.t. and stirred for additional 21 h. It was poured into ice water (30 ml) and washed with brine (2 x 30 ml). The brine was extracted with CH₂Cl₂ (50 ml), the combined org. soln. dried (Na₂SO₄) and the solvent removed. The residue was flash chromatographed (120 g silica gel; hexane/ ethylacetate 6:4) and bulb-to-bulb distilled (130° (oven temp.)/2·10⁻² mbar) to give **4** (2.35 g, 88%) as a practically colorless, lachrymatory liquid (1:1 mixture of diastereomers as determined by GC and 400 MHz ¹H-NMR). ¹H-NMR (60 MHz, CDCl₃): 1.25 (*t*, *J* = 7, ethoxy-CH₃); 1.80 (*ddd*, *J* = 15 and 5 and 2, HC(5)); 2.06 (*s*, acetate CH₃); 2.87 (*m*, HC(5)); 3.40 (*d*, *J* = 6, CH₂Br); 3.70 (*m*, ethoxy-CH₂); 4.78 (*m*, HC(4), OCHO); 5.53 (*m*, HC(1)); 6.10 (AB-system, HC(2), HC(3)). IR (neat): 2980*m*, 2930*w*, 2900*w*, 1740*s*, 1430*w*, 1365*m*, 1240*s*, 1190*w*, 1115*m*, 1060*m*, 1020*s*, 910*w*, 770*w*. MS (20°): a.o. 169(5), 167(5), 153(33), 151(33), 126(7), 125(91), 123(30), 83(31), 82(21), 81(5), 73(7), 72(9), 43(100).

(1*R*,4*S*)-4-(1-Ethoxy-2-bromoethoxy)-2-cyclopentene-1-ol (**5**). A solution of bromoacetal **4** (1.125 g, 3.8 mmol) in 0.25M KOH in 90% ethanol (100 ml) was stirred in the dark at r.t. for 1.5 h. It was neutralized (pH = 7 to 8) by dropwise addition of 98% acetic acid. After concentrating *in vacuo* to ca. 10 ml, the resulting two phases were poured into ice water (40 ml) and extracted with Et₂O (3 x 50 ml). The org. soln. was washed with brine (40 ml), dried (Na₂SO₄) and the solvent evaporated. The residue was flash chromatographed (100 g silica gel; hexane/ethylacetate 6:4) and bulb-to-bulb distilled (130° (oven temp.)/2·10⁻² mbar) to give **5** (0.798 g, 83%) as a pale yellow liquid. ¹H-NMR (60 MHz, CDCl₃): 1.23 (*t*, *J* = 7.5, ethoxy-CH₃); 1.70 (*ddd*, *J* = 14.5 and 4 and 1.5, HC(5)); 2.35 (*s*, OH); 2.75 (*md*, *J* = 2, HC(5)); 3.37 (*d*, *J* = 5.5, CH₂Br); 3.40 - 3.95 (*m*, ethoxy-CH₂); 4.53 - 4.80 (*m*, HC(4), OCHO); 4.83 (*t*, *J* = 5.5, HC(1)); 6.07 (AB-system, HC(2), HC(3)). IR (neat): 3410*m*, 3060*w*, 2980*m*, 2950*m*, 2890*m*, 1445*w*, 1425*m*, 1370*m*, 1355*m*, 1320*w*, 1190*m*, 1120*s*, 1060*s*, 1030*s*, 890*w*, 765*w*. MS (20°): a.o. 154(17), 152(17), 126(19), 124(20), 85(6), 84(100), 83(6), 76(6), 74(13), 73(8), 56(13), 48(7), 46(12), 44(10), 29(10).

(1,1-Dimethylethyl){[(1*R*,4*S*)-4-(1-ethoxy-2-bromoethoxy)-2-cyclopentene-1-yl]oxy}dimethylsilane (**6**). To a solution of *tert*-butyldimethylchlorosilane (0.33 g, 2.2 mmol) and imidazole (0.32 g, 4.6 mmol) in DMF (2 ml) under N₂ was added bromoacetal **5** (0.465 g, 1.9 mmol). After stirring at r.t. for 1.5 h the clear solution was poured into ice water (50 ml) and extracted with Et₂O (3 x 30 ml). The org. soln. was washed with 0.1M P-buffer (pH = 6; 2 x 30 ml) and H₂O (30 ml), dried (Na₂SO₄) and the solvent removed. Bulb-to-bulb distillation (30 min at 60° (oven temp.)/ 2·10⁻² mbar to remove volatile impurities, then 140°/2·10⁻²mbar) afforded **6** (0.661 g, 98%) as a colorless and odorless liquid. ¹H-NMR (60 MHz, CDCl₃): 0.08 (*s*, SiCH₃); 0.91 (*s*, Si-*tert*-butyl); 1.25 (*t*, *J* = 7, ethoxy-CH₃); 1.40 - 1.95 (*m*, HC(5')); 2.50 - 3.00 (*m*, HC(5')); 3.38 (*d*, *J* = 5.5, CH₂Br); 3.47 - 3.94 (*m*,

ethoxy-CH₂); 4.50 - 4.95 (*m*, HC(1'), HC(4'), OCHO); 5.95 (AB-system, HC(2'), HC(3')). IR (neat): 2960*m*, 2930*s*, 2890*m*, 2860*m*, 1470*w*, 1460*w*, 1370*m*, 1250*m*, 1125*s*, 1100*s*, 1060*s*, 1030*s*, 1005*m*, 940*w*, 905*m*, 835*s*, 775*m*, 670*w*. MS (20°): a.o. 263(5), 261(5), 198(5), 197(31), 185(23), 183(21), 181(21), 157(44), 153(94), 151(100), 141(15), 125(66), 123(71), 115(9), 104(8), 103(79), 75(90), 73(91), 72(25), 66(13), 59(14), 57(21), 44(9), 43(9). **6** can also be prepared from **3** in 96% overall yield without chromatography of **4** and **5** respectively. The only purification is the bulb-to-bulb distillation of **6**.

1-Octyne-3-one (7). To an ice cooled solution of 1-octyne-3-ol (8.00 g, 63.4 mmol) in distilled acetone (30 ml) was added a solution of CrO₃ (5.49 g, 54.9 mmol) in 95% H₂SO₄ (5 ml) and H₂O (15.7 ml) with the temp. not exceeding 10°. After stirring at r.t. for 1 h H₂O (30 ml) was added to dissolve the precipitate. The resulting two liquid phases were poured into ice water (140 ml) and extracted with Et₂O (3 x 50 ml and 2 x 25 ml). The org. soln. was washed with 5% NaHCO₃ (50 ml) and brine (3 x 50 ml), dried (MgSO₄) and the solvent removed. Distillation (60° - 62°/ca. 10 mbar) afforded **7** (7.26 g, 92%) as a pale yellow liquid with an intense, characteristic odor. ¹H-NMR (60 MHz, CDCl₃): 0.69 - 2.05 (*m*, H on C(5) to C(8)); 2.60 (*t*, *J* = 7, HC(4)); 3.24 (*s*, HC(1)). IR (neat): 3260*m*, 2960*m*, 2930*m*, 2880*m*, 2090*m*, 1680*s*, 1460*w*, 1405*w*, 1380*w*, 1220*w*, 1130*w*, 1080*w*, 700*w*. n_D²⁰: 1.4330.

*B*₁₂-catalyzed electrosynthesis of the prostaglandin precursors **8**:

trans-1-[(2*R*/*S*,3*aR*,4*R*,5*R*,6*aS*)-5-[[[1,1-Dimethylethyl]dimethylsilyl]oxy]-2-ethoxyhexahydro-2H-cyclopenta[b]furan-4-yl]-1-octene-3-one (**8a,b**). The cathode compartment of the electrochemical cell, containing C-felt (2.77 g) as cathode material, was charged with a solution of vitamin B_{12a} (250 mg, 0.18 mmol) in 0.3M LiClO₄/DMF (250 ml). The anode compartment was charged with 0.3M LiClO₄/DMF (ca. 50 ml). B₁₂ was reduced to Co(I) at a constant cathode potential of -1.4 V (vs. SCE) until the initial current of ca. 50 mA had diminished to a stable background level and the color had changed from red to dark green. On reducing the potential to -0.9 V a stable background level of ca. 3.5 mA was observed. Keeping the potential at -0.9 V, 1-octyne-3-one **7** (3.72 g, 30 mmol) was added, followed by bromoacetal **6** (3.65 g, 10 mmol) after 15 min. Thereby the color changed to dark red and the current remained on a low level. After switching on the 250 W halogen lamp the current rose to 30 mA and the temperature to 40° - 41°, which was maintained throughout the reaction. Additional **7** was added after 5 h (2.48 g, 20 mmol) and 25 h (1.24 g, 10 mmol). After 44 h the current had diminished to a constant background level of 9 mA and 1543 Cb e⁻ (16.0 mmol) had been consumed. The solution was poured into ice water (750 ml) and extracted with Et₂O (6 x 100 ml). The org. soln. was washed with brine (3 x 100 ml), dried (Na₂SO₄) and the solvent removed. Excess **7** (ca. 1.8 g) was removed by bulb-to-bulb distillation (45° (oven temp.)/5·10⁻² mbar) and the residual dark brown liquid (8.86 g) was filtered through silica gel (70 g; low boiling petroleum ether/Et₂O 8:2; an attempt to accomplish *cis-trans*-isomerization at this stage with a trace of I₂ in hexane and irradiating with sunlight was unsuccessful). Flash chromatography of the crude product (210 g silica gel; low boiling petroleum ether/Et₂O 19:1 to 8:2) afforded two main fractions: 1,3,5-tri(hexane-1-one-1-yl)benzene (0.30 g, 0.8 mmol; data see below) and 3.50 g of a yellow liquid comprising the four isomers of **8** (containing at least 65% **8** (GC), corresponding to at least 55% with respect to **6**). For *cis-trans*-isomerization the latter fraction was dissolved in hexane (50 ml) containing I₂ (90 mg) and irradiated with a halogen lamp for 50 min at r.t. The solution was then poured into ice water (25 ml), washed with brine, containing a small amount of Na-thiosulfate (2 x 25 ml), and brine (25 ml), dried (Na₂SO₄) and the solvent removed. Flash chromatography (150 g silica gel; low boiling petroleum ether/Et₂O 19:1 to 8:2) of the remaining dark liquid (3.36 g) afforded essentially pure **8a** (1.07 g, 26%) and **8b** (0.85 g, 21%) as slightly yellow oils (assignment of configuration based on NOE-experiments; each of the pure diastereomers **8a** and **8b** yielded an equilibrium mixture of both in ethanol containing a trace amount of

p-toluenesulfonic acid monohydrate). For elemental analysis a sample of **8a** was distilled in a glass tube at 150° (oven temp.)/ $5 \cdot 10^{-3}$ mbar. $C_{23}H_{42}O_4Si$: calculated C 67.27%, H 10.31%; found C 67.27%, H 10.23%.

Data of **8a** (R_f -value = 0.51 in low boiling petroleum ether/Et₂O 8:2): ¹H-NMR (400 MHz, CDCl₃): -0.04 (*s*, Si-CH₃); -0.02 (*s*, Si-CH₃); 0.82 (*s*, Si-*tert.*-butyl); 0.70 - 0.95 (*m*, HC(8)), 1.15 (*td*, $J = 7.07$ and 0.77, ethoxy-CH₃); 1.10 - 1.35 (*m*, HC(6), HC(7)); 1.50 - 1.65 (*m*, HC(5)); 1.72 (*ddd*, $J = 13.48$ and 8.86 and 4.34, HC(6'')); 1.84 (*ddd*, $J = 13.48$ and 4.83 and 4.83, HC(3'')); 2.01 (*ddd*, $J = 13.52$ and 8.82 and <1, HC(3'')), 2.25 - 2.45 (*m*, HC(3a'), HC(4'), HC(6'')); 2.48 (*t*, $J = 7.45$, HC(4)); 3.34 - 3.48 (*m*, ethoxy-CH₂); 3.67 (*dqd*, $J = 9.61$ and 7.11 and 0.76, ethoxy-CH₂); 3.89 (*ddd*, $J = 8.15$ and 8.15 and 8.15, HC(5'')); 4.42 (*ddd*, $J = 7.02$ and 7.02 and 4.20, HC(6a'')); 5.19 (*d*, $J = 4.91$, HC(2'')); 6.09 (*dd*, $J = 15.84$ and 0.96, HC(2)); 6.64 (*dd*, $J = 15.83$ and 8.05, HC(1)). ¹³C-NMR (25.2 MHz, CDCl₃): -4.7u, -4.5u, 13.9u, 15.2u, 18.0g, 22.5g, 24.0g, 25.7u, 31.5g, 38.4g, 40.4g, 41.0g, 44.3u, 57.0u, 62.6g, 78.1u, 79.5u, 105.1u, 130.8u, 146.2u. IR (neat): 3450w, 2960s, 2930s, 2860m, 2250w, 1695m, 1680m, 1630w, 1465m, 1405w, 1380m, 1360w, 1335w, 1250m, 1115m, 1050m, 1005m, 910m, 865w, 835m, 775m, 730s, 670w, 645w. MS (90°): a.o. 365(10), 354(11), 353(43), 309(11), 308(12), 307(50), 281(17), 234(15), 233(70), 199(12), 189(13), 187(23), 178(11), 161(14), 153(12), 151(13), 136(14), 135(11), 133(14), 131(11), 129(11), 117(29), 107(18), 105(10), 101(23), 99(100), 91(15), 81(14), 79(15), 75(63), 73(62), 72(21), 71(48), 59(14), 55(12), 44(11), 43(49), 41(11). UV (CCl₄): 266(3.29).

Data of **8b** ($R_f = 0.29$): ¹H-NMR (400 MHz, CDCl₃): -0.03 (*s*, Si-CH₃); -0.01 (*s*, Si-CH₃); 0.83 (*s*, Si-*tert.*-butyl); 0.75 - 0.98 (*m*, HC(8)); 1.18 (*t*, $J = 7.10$, ethoxy-CH₃); 1.15 - 1.40 (*m*, HC(6), HC(7)); 1.52 - 1.64 (*m*, HC(5)); 1.76 (*ddd*, $J = 12.61$ and 10.73 and 6.36, HC(6'')); 1.84 (*d*, $J = 13.32$, HC(3'')); 1.99 (*ddd*, $J = 13.36$ and 9.36 and 5.29, HC(3'')); 2.30 - 2.40 (*m*, HC(3a'')); 2.37 - 2.45 (*m*, HC(6'')); 2.49 (*t*, $J = 7.47$, HC(4)); 2.86 (*ddd*, $J = 9.23$ and 9.23 and 9.23, HC(4'')); 3.40 (*dq*, $J = 9.55$ and 7.03, ethoxy-CH₂); 3.71 - 3.80 (*m*, ethoxy-CH₂, HC(5'')); 4.49 (*dd*, $J = 14.35$ and 7.70, HC(6a'')); 5.15 (*d*, $J = 5.20$, HC(2'')); 6.17 (*dd*, $J = 15.83$ and 0.99, HC(2)); 6.69 (*dd*, $J = 15.83$ and 8.10, HC(1)). ¹³C-NMR (25.2 MHz, CDCl₃): -4.6u, -4.4u, 13.9u, 15.1u, 18.0g, 22.5g, 24.0g, 25.7u, 31.5g, 37.1g, 40.4g, 43.4g, 44.0u, 55.5u, 62.5g, 76.8u, 81.0u, 105.5u, 130.8u, 147.1u. IR (neat): 3460w, 2960s, 2930s, 2860m, 2250w, 1695m, 1675m, 1630m, 1465m, 1410w, 1375m, 1255m, 1120m, 1055m, 1005m, 980m, 910m, 865m, 835m, 775m, 730s, 665w, 645w. MS (80°): a.o. 365(4), 353(6), 309(18), 308(24), 307(100), 234(10), 233(44), 189(11), 187(24), 133(14), 117(20), 107(11), 103(11), 101(13), 99(62), 91(10), 81(12), 79(10), 75(41), 73(41), 72(14), 71(30), 59(10), 43(34). UV (CCl₄): 268(3.41).

Data of *1,3,5-tri(hexane-1-one-1-yl)benzene*: ¹H-NMR (60 MHz, CDCl₃): 0.70 - 1.20 (*m*, HC(6'')); 1.20 - 2.20 (*m*, H on C(3') to C(5'')); 3.10 (*t*, $J = 7$, HC(2'')); 8.76 (*s*, aromatic H). IR (neat): 2860s, 2830s, 2770s, 2760s, 1695s, 1595w, 1470m, 1460m, 1440w, 1410w, 1380m, 1315w, 1295w, 1240m, 1170s, 1115w, 1055w, 915m, 840w, 775w, 730s, 680w, 645w. MS (90°): a.o. 372(26), 317(7), 316(27), 302(13), 301(61), 272(11), 261(17), 260(89), 245(23), 217(12), 205(13), 204(100), 203(11), 99(33), 71(24), 43(17). UV (EtOH): 228(4.37), 300(3.19).

cis-1-[(2*R*/*S*,3*aR*,4*R*,5*R*,6*aS*)-5-[(1,1-Dimethylethyl)dimethylsilyloxy]-2-ethoxyhexahydro-2*H*-cyclopenta[*b*]-furan-4-yl]-1-octene-3-one (**8c,d**). In a parallel experiment under the same conditions, but without *cis*-*trans*-isomerization, all four isomers of **8** were separated by flash chromatography (320 g silica gel; low boiling petroleum ether/Et₂O 19:1 to 8:2) affording **8a** (0.54 g, $R_f = 0.51$ in low boiling petroleum ether/Et₂O 8:2), **8b** (0.51 g, $R_f = 0.29$), **8c** (0.34 g, $R_f = 0.66$) and **8d** (0.52 g, $R_f = 0.37$).

Data of **8c** ($R_f = 0.66$): ¹H-NMR (400 MHz, CDCl₃): *ca.* 0 (*s*, Si-CH₃); 0.83 (*s*, Si-*tert.*-butyl); 0.85 - 1.00 (*m*, HC(8)); 1.19 (*t*, $J = 7.08$, ethoxy-CH₃); 1.21 - 1.42 (*m*, HC(6), HC(7)); 1.55 - 1.63 (*m*, HC(5)); 1.74 - 1.81 (*m*, HC(3'), HC(6'')); 1.97 (*ddd*, $J = 13.3$ and 9.0 and 1.6, HC(3'')); 2.21 - 2.40 (*m*, HC(3a'), HC(4'), HC(6'')); 2.45 (*t*, $J = 7.43$, HC(4)); 3.44 (*dq*, $J = 9.61$ and 7.04, ethoxy-CH₂); 3.72 (*dq*, $J = 9.66$ and 7.11, ethoxy-CH₂); 3.87 (*ddd*, J

= 8.29 and 8.29 and 6.95, HC(5'')); 4.47 (*ddd*, $J = 7.05$ and 7.05 and 4.24 , HC(6a'')); 5.29 (*dd*, $J = 5.20$ and 1.59 , HC(2'')); 5.72 (*dd*, $J = 11.44$ and 10.53 , HC(1)); 6.21 (*dd*, $J = 11.47$ and 0.69 , HC(2)). IR (neat): 3450w, 2960s, 2930s, 2860s, 1695s, 1625w, 1465m, 1410w, 1375w, 1335w, 1295w, 1250m, 1130m, 1110m, 1055m, 1030m, 1000m, 940w, 905w, 835m, 775m, 730w, 670w. MS (65°): a.o. 365(9), 354(27), 353(100), 308(20), 307(83), 281(27), 279(10), 261(16), 234(13), 233(58), 215(10), 206(12), 199(11), 189(16), 187(35), 181(10), 178(11), 151(17), 153(12), 151(37), 136(14), 133(15), 131(14), 117(27), 107(20), 101(22), 99(94), 91(16), 81(14), 79(13), 75(70), 73(55), 71(52), 59(14), 55(12), 43(52), 41(12), 18(12).

Data of **8d** ($R_f = 0.37$): ¹H-NMR (400 MHz, CDCl₃): ca. 0 (*s*, Si-CH₃); 0.84 (*s*, Si-*tert*-butyl); 0.85 - 1.00 (*m*, HC(8)); 1.22 (*t*, $J = 7.04$, ethoxy-CH₃); 1.25 - 1.40 (*m*, HC(6), HC(7)); 1.50 - 1.70 (*m*, HC(5)); 1.83 (*ddd*, $J = 12.48$ and 10.65 and 6.49 , HC(6'')); 1.96 (*ddd*, $J = 13.31$ and 9.36 and 5.20 , HC(3'')); 2.08 (*d*, $J = 13.27$, HC(3'')); 2.20 - 2.80 (*m*, HC(4), HC(3a'), HC(4'), HC(6'')); 3.49 (*q*, $J = 7.02$, ethoxy-CH₂); 3.70 - 3.90 (*m*, HC(5'')); 4.52 (*dd*, $J = 14.82$ and 7.57 , HC(6a'')); 5.19 (*d*, $J = 5.16$, HC(2'')); 5.74 (*dd*, $J = 10.82$ and 11.77 , HC(1)); 6.13 (*dd*, $J = 11.83$ and 0.59 , HC(2)). IR (neat): 3450w, 2960s, 2930s, 2860s, 1715m, 1680m, 1625w, 1465m, 1410w, 1380m, 1250m, 1180w, 1120s, 1055s, 1005m, 980w, 910w, 870w, 835m, 775m, 730m, 670w, 645w. MS (70°): a.o. 365(2), 364(2), 353(5), 309(17), 308(24), 307(100), 233(18), 189(9), 187(26), 151(11), 134(6), 117(13), 101(7), 99(38), 81(6), 75(29), 73(25), 72(10), 71(24), 43(31).

(3*aR*,6*aS*)-2-Ethoxy-3,3*a*,6,6*a*-tetrahydro-2H-cyclopenta[b]furan (**9a,b**). A suspension of vitamin B_{12a} (55 mg, 0.04 mmol) and NH₄Cl (3.21 g, 60 mmol) in ethanol (30 ml) was stirred at r.t. under Ar with activated Zn-wool (3.00 g, 46 mmol) wrapped around the stirrer bar. After 10 to 20 min the color of the solution changed from red to dark green. Bromoacetal **4** (1.173 g, 4 mmol) was added and the mixture stirred at r.t. for 6 h. Then the red-brown suspension was poured into a solution of 25% NH₃ (5 ml) in ice water (150 ml) and extracted with Et₂O (5 x 30 ml). The org. soln. was washed with brine (5 x 30 ml), dried (Na₂SO₄) and the solvent removed. Bulb-to-bulb distillation (r.t./5·10⁻² mbar) afforded **9** (0.453 g, 73%) as a clear liquid with a characteristic odor (1:1 mixture of diastereomers **9a** and **9b** as determined by GC). For elemental analysis a sample was purified by preparative GC. C₉H₁₄O₂: calculated C 70.10%, H 9.15%; found C 69.86%, H 9.12%. IR (neat): 3060w, 2980m, 2920m, 2870m, 1445w, 1370w, 1330m, 1190w, 1170w, 1120m, 1100m, 1060s, 1040s, 1005m, 990s, 865m, 820w. MS (20°): a.o. 154(8), 126(6), 125(5), 110(9), 109(60), 108(68), 98(10), 88(10), 83(11), 81(66), 80(67), 79(100), 77(17), 73(9), 72(67), 70(8), 67(22), 66(20), 53(10), 44(28), 43(22), 41(20), 39(9), 29(8), 18(10). For ¹H-NMR the two diastereomers were separated by flash chromatography (silica gel; hexane/ethylacetate 9:1). ¹H-NMR (60 MHz, CDCl₃): 1. diastereomer ($R_f = 0.53$): 1.20 (*t*, $J = 7$, ethoxy-CH₃); 1.57 - 2.45 (*m*, HC(6)); 2.57 (*m*, HC(3)); 3.19 - 4.10 (*m*, HC(3*a*), ethoxy-CH₂); 4.78 (*m*, HC(6*a*)); 5.18 (*dd*, $J = 4.5$ and 2 , OCHO); 5.63 (AB-system, HC(4), HC(5)); 2. diastereomer ($R_f = 0.42$): 1.12 (*t*, $J = 7$, ethoxy-CH₃); 1.65 - 2.45 (*m*, HC(6)); 2.61 (*m*, HC(3)); 3.13 - 4.00 (*m*, HC(3*a*), ethoxy-CH₂); 4.87 (*m*, HC(6*a*)); 5.14 (*dd*, $J = 4.5$ and 2 , OCHO); 5.67 (AB-system, HC(4), HC(5)).

(3*aR*,6*aS*)-3,3*a*,6,6*a*-Tetrahydro-2H-cyclopenta[b]furan-2-ol (**10**). A solution of acetal **9** (0.326 g, 2.1 mmol) in 1.5% HCl/THF (3:2; 20 ml) was stirred at r.t. for 48 h. It was poured into ice water (80 ml) and extracted with Et₂O (5 x 25 ml). The org. soln. was washed with 5% NaHCO₃ (25 ml) and brine (3 x 25 ml), dried (Na₂SO₄) and the solvent removed. Recrystallization from low boiling petroleum ether afforded **10** (0.098 g, 37%) as colorless crystals. M.p. 54° - 56°. ¹H-NMR (60 MHz, CDCl₃): 1.60 - 2.40 (*m*, HC(6)); 2.30 - 2.90 (*m*, HC(3)); 3.20 - 3.77 (*m*, HC(3*a*)); 4.37 (*s*, OH); 4.91 (*m*, HC(6*a*)); 5.32 - 5.90 (*m*, OCHO, HC(4), HC(5)). IR (KBr): 3380s, 3050m, 3000m, 2950m, 2930m, 2850m, 1445w, 1430w, 1360w, 1330w, 1310w, 1285w, 1265m, 1250m, 1220w, 1190m, 1175m, 1110m, 1090m, 1060m, 1035s, 995m, 975s, 920w, 890m, 860m, 820m, 770w, 710m. MS: a.o. 126(5),

108(37), 83(47), 81(11), 80(39), 79(100), 77(26), 70(8), 67(30), 66(36), 65(8), 55(22), 54(13), 53(16), 51(11), 44(8), 43(11), 41(21), 39(22), 28(13), 27(14), 18(24). C₇H₁₀O₂: calculated C 66.65%, H 7.99%; found C 66.23%, H 7.96%.

(-)-(3*aR*,6*aS*)-3,3*a*,6,6*a*-Tetrahydro-2H-cyclopenta[*b*]furan-2-on ((-)-**11**). To an ice cooled solution of lactol **10** (0.082g, 0.65 mmol) in distilled acetone (2 ml) was added a solution of CrO₃ (0.048 g, 0.48 mmol) and 95% H₂SO₄ (0.04 ml) in H₂O (0.14 ml). After stirring at r.t. for 2 h H₂O (3 ml) was added to dissolve the precipitate and the solution poured into ice water (40 ml) and extracted with Et₂O (3 x 20 ml). The org. soln. was washed with 5% NaHCO₃ (20 ml) and brine (3 x 20 ml), dried (Na₂SO₄) and the solvent removed. The residue was sublimed in a bulb-to-bulb distillation apparatus (70°(oven temp.)/4·10⁻² mbar) affording (-)-**11** (0.051 g, 63%) as colorless crystals. M.p. 44° - 45°. ¹H-NMR (60 MHz, CDCl₃): 2.20 - 3.04 (*m*, HC(3), HC(6)); 3.30 - 3.80 (*m*, HC(3*a*)); 5.18 (*ddd*, *J* = 6 and 3 and 3, HC(6*a*)); 5.45 - 6.03 (*m*, HC(4), HC(5)). IR (KBr): 3500*w*, 3080*w*, 3060*w*, 3000*m*, 2960*w*, 2930*m*, 2910*m*, 2830*w*, 1765*s*, 1430*m*, 1420*m*, 1365*m*, 1350*m*, 1295*m*, 1255*s*, 1230*m*, 1190*s*, 1170*s*, 1150*s*, 1110*w*, 1040*s*, 1015*m*, 955*m*, 940*m*, 920*s*, 895*s*, 865*m*, 825*m*, 805*m*, 775*m*, 730*s*. MS (250°): a.o. 125(3), 124(29), 96(39), 95(83), 91(9), 81(43), 79(27), 77(18), 68(22), 67(100), 66(10), 65(19), 54(29), 53(25), 52(9), 51(17), 50(16), 42(10), 41(39), 40(13), 39(65), 38(13), 37(10), 32(18), 28(51), 27(21), 18(84). [α]_D²⁵: -103° (*c* = 0.80, CH₃OH; Lit.value²⁷): [α]_D²⁵: -106° (*c* = 0.80, CH₃OH)). C₇H₈O₂: calculated C 67.73%, H 6.50%; found C 67.45%, H 6.51%.

(+)-(R)-2-Cyclopentene-1-ol (**13**). A 100 ml flask under Ar, containing MeOH (35 ml), was charged with vitamin B_{12a} (1.29 g, 0.83 mmol), NH₄Cl (1.0 g) and Zn-wool (1.5 g) wrapped around a magnetic stirrer bar. After stirring for 30 min the color changed from red (B_{12a}) to dark green (B_{12s}). To this mixture was added cyclopentene oxide **12** (6.96 g, 82.7 mmol) *via* syringe. The color immediately changed to red (alkyl-Co(III)). After stirring for 8 days at r.t. (*ca.* 22°) under Ar, the solution had turned dark brown. The flask was opened, the stirrer bar (together with excess Zn-wool) removed and Et₂O (100 ml) added. A red-brown precipitate (B₁₂) formed. The solution was decanted and the precipitate extracted with Et₂O (5 x 50 ml). The combined ethereal fractions were washed with brine (2 x 60 ml), dried (Na₂SO₄) and the solvent evaporated. The remaining oil was bulb-to-bulb distilled (30° (oven temp.)/*ca.* 2·10⁻² mbar) affording **13** (5.01 g of a purity of 93% (GC), corresponding to 67% with respect to **12**) as a colorless oil. [α]_D²²: +91.0° (neat), corresponding to 62% e.e.²⁹).

1-(1-Ethoxy-2-bromoethoxy)-2-cyclopentene (**14**). A suspension of crude cyclopentenol **13** (5.01 g, 59.5 mmol, [α]_D²²: +91.0°, e.e.=62%) and N-bromosuccinimide (10.8 g, 65.0 mmol) in CH₂Cl₂ (80 ml) was cooled to -50° under Ar. Ethylvinylether (6.43 g, 89.3 mmol) was added dropwise over a period of 20 min. The mixture was stirred for 2 h at -40° to -50° and then at r.t. for 16 h. The suspension was poured into ice water (100 ml) and extracted with CH₂Cl₂ (6 x 100 ml). The combined organic fractions were washed with brine (3 x 50 ml), dried (Na₂SO₄) and the solvent removed. Bulb-to-bulb distillation (95° (oven temp.)/*ca.* 2·10⁻² mbar) of the residue afforded bromoacetal **14** (15.0 g of a purity of 93% (GC), corresponding to 100% with respect to **13**) as a colorless oil. ¹H-NMR (60 MHz, CDCl₃): 1.23 (*t*, 3H), 1.70 - 2.80 (*m*, 4H), 3.20 - 3.90 (*m*, 4H), 4.60 - 5.00 (*m*, 2H), 5.60 - 6.20 (*m*, 2H). IR (neat): 3060*w*, 2980*s*, 2930*m*, 2900*m*, 2860*m*, 1725*w*, 1620*w*, 1485*w*, 1425*w*, 1360*w*, 1325*w*, 1190*w*, 1120*s*, 1040*s*, 920*w*. MS: 153(12), 151(12), 125(17), 123(17), 109(5), 103(12), 84(6), 83(4), 77(5), 67(100).

(3*aS*,6*aR*)-2-Ethoxy-3,3*a*,6,6*a*-tetrahydro-2H-cyclopenta[*b*]furan (**15a,b**). To a solution of crude bromoacetal **14** (0.94 g, 4.0 mmol) in ethanol (20 ml) was added vitamin B_{12a} (0.20 g, 0.14 mmol). The solution was heated to 60° under Ar and powdered sodium borohydride (0.23 g, 6.0 mmol) added batchwise over a period of 1 h. The

temperature of the reaction was kept at 60° to 65°. After additional 10 min the mixture was allowed to cool to r.t. and ice water (20 ml) was added. The mixture was extracted with Et₂O (6 x 40 ml). The ethereal fraction was washed with 2N NaHSO₄ (20 ml), brine (2 x 60 ml), dried (Na₂SO₄) and the solvent removed *in vacuo* affording **15** (0.51 g of a purity of 85% (GC), corresponding to 71% with respect to **14** as a 1:1 mixture of diastereomers **15a** and **15b** (GC)) as a yellowish oil. IR (neat): 3060w, 2980m, 2920m, 2870m, 1445w, 1370w, 1330m, 1190w, 1170w, 1120m, 1100m, 1060s, 1040s, 1005m, 990s, 865m, 820w. MS: 154(8), 110(9), 109(60), 108(68), 98(10), 88(10), 83(11), 81(66), 80(67), 79(100), 77(17), 73(9), 72(67), 70(8), 67(22), 66(20), 53(10), 44(28), 43(22), 41(20), 39(9), 29(8), 18(10). For ¹H-NMR the two diastereomers were separated by flash chromatography (silicagel; Et₂O/hexane 1:1). ¹H-NMR (60 MHz, CDCl₃): 1. diastereomer (R_f = 0.63 in Et₂O/hexane): 1.20 (t, J = 7, ethoxy-CH₃); 1.5 - 2.4 (m, HC(6)); 2.57 (m, HC(3)); 3.1 - 4.1 (m, HC(3a), ethoxy-CH₂); 4.78 (m, HC(6a)); 5.16 (dd, J = 4.5 and 2, OCHO); 5.63 (AB-system, HC(4), HC(5)); 2. diastereomer (R_f = 0.56): 1.12 (t, J = 7, ethoxy-CH₃); 1.6 - 2.5 (m, HC(6)); 2.61 (m, HC(3)); 3.1 - 4.0 (m, HC(3a), ethoxy-CH₂); 4.87 (m, HC(6a)); 5.14 (dd, J = 4.5 and 2, OCHO); 5.67 (AB-system, HC(4), HC(5))

(+)-(3a*S*,6a*R*)-3,3a,6,6a-Tetrahydro-2H-cyclopenta[b]furan-2-on ((+)-**11**). To an ice cooled solution of crude acetal **15** (2.46 g, 16.0 mmol) in distilled acetone (90 ml) was added dropwise a solution of CrO₃ (3.2 g, 32 mmol) in 95% H₂SO₄ (3.37 ml) dissolved in H₂O (13 ml) over a period of 60 min. After stirring at 0° for 1 h and at r.t. for 2 h, ice water (50 ml) was added to dissolve the precipitate. The solution was poured into ice water (50 ml) and extracted with Et₂O (6 x 150 ml). The combined organic solution was washed with 5% NaHCO₃ (2 x 60 ml), brine (3 x 100 ml), dried (Na₂SO₄) and the solvent removed *in vacuo*. Bulb-to-bulb distillation (70° (oven temp.)/*ca.* 4·10⁻² mbar) afforded (+)-**11** (1.67 g of a purity of 93% (GC), corresponding to 78% with respect to **15**) as a colorless oil ([α]_D: +91.0°, 62% e.e.). For crystallization the oil was dissolved in Et₂O/hexane *ca.* 1:1 (*ca.* 40 ml) and cooled gradually to 10° (20 min), 0° (2 h), -20° (24 h), affording colorless crystals which were recrystallized twice from Et₂O/hexane giving pure (+)-**11** (0.89 g, 45%). M.p. 44.5 - 45.5°. [α]_D²²: +104.4° (c = 0.802, CH₃OH). The e.e. was found to be >99.5% by analytical GC with heptakis(2,3,6-tri-O-pentyl)-β-cyclodextrine as chiral stationary phase. ¹H-NMR, IR and MS of (+)-**11** are virtually superimposable to those of (-)-**11** described above. C₇H₈O₂: calculated C 67.73%, H 6.50%; found C 67.45%, H 6.59%.

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